Chapter 2 **Electronics I - Semiconductors** Fall 2017

Charged Particles

- The operation of all electronic devices is based on controlling the flow of charged particles
- There are two type of charge in solids
 - Electrons
 - Holes
- There are two mechanism through which charge can be transported in a material
 - Drift (motion of charge caused by an electric field)
 - Diffusion (motion resulting from a non-uniform charge distribution)

Electronic Structure of the elements

- Atom's chemical activity depends on the electrons in the outermost shells (orbits).
 These electrons are called VALENCE electrons.
- In extremely pure elements, such as silicon, the atoms arrange themselves in regular patterns called CRYSTALS. The valence electrons determine the exact shape (= LATTICE structure)

Electronic orbitals in silicon

Element	Atomic Number	Configuration
Si	14	(1s) ² (2s) ² (2p) ⁶ (3s) ² (3p) ²
s, p, d designate orbital shap s holds up to 2 electrons	e $(1s)^2$ (filled)	
p holds up to 6 electrons d holds up to 10 electrons	$(2s)^2$ (filled)	
	(2p) ⁶ (filled)	electronic orbitals in silicon
	(3sp) ⁴	
Source: Howe & Sodini	4 tetrahedral hybrid orbitals, half-filled with 4 valence electrons	4

Silicon Crystal Lattice



where n_{cell} is the number of unit cells per cm³. Since each atom has four tetrahedral bonds and contributes one electron to each bond, the concentration of valence electrons is $4 N_{Si} = 2 \times 10^{23} \text{ cm}^{-3}$.

2D-Representation of Silicon Crystal



Figure 3.1 Two-dimensional representation of the silicon crystal. The circles represent the inner core of silicon atoms, with +4 indicating its positive charge of +4q, which is neutralized by the charge of the four valence electrons. Observe how the covalent bonds are formed by sharing of the valence electrons.

At 0 K, all bonds are intact and no free electrons are available for current conduction.

Free electrons and holes



Figure 3.2 At room temperature, some of the covalent bonds are broken. Each broken bond gives rise to a free electron and a hole, both of which become available for current conduction.

Periodic Table

Table 1.2	Abbreviated Periodic Chart of the Elements.				
Ш	III	IV	v	VI	
4	5	6	7	8	
Be	B	C	N	0	
12	13	14	15	16	
Mg	Al	Si	P	S	
30	31	32	33	34	
Zn	Ga	Ge	As	Se	
48	49	50	51	52	
Cd	In	Sn	Sb	Te	
80	81	82	83	84	
Hg	Tl	Pb	Bi	Po	

Energy Band Structure

- Band gap energy (E_g) is the minimum energy to dislodge an electron from its covalent bond.
- For Silicon at room temp. (T=300 °K) $E_g = 1.12 \text{ eV} = 1.792 \times 10^{-19} \text{ Joule}$



Source: Millman & Halkias

Concentration of free electrons

• The concentration of electrons (and holes) in pure silicon at room temperature is approximately:

 $n_i(T = 300 \ K) \cong 1 \times 10^{10} \ cm^{-3}$

 As temperature increases, the intrinsic concentration n_i approximately doubles every 10°C rise over room temperature (source: Howe & Sodini)

$$n_i(T) \cong n_i(T_{300}) \times 2^{\frac{T - T_{300}}{10}} cm^{-3}$$

- Given that the number of bonds is 2×10²³ cm⁻³, at room temperature only an extremely small fraction of the bonds are broken (1 in 20×10¹² bonds, that is 1 in 5×10¹² atoms)
- Too few: we need more !!!

Intrinsic carrier concentration as a function of temperature



E_G also depends on T

$$n_i = \sqrt{A_C A_V} T^{3/2} e^{-E_G/(2KT)} \cong 5 \times 10^{15} T^{3/2} e^{-E_G/(2KT)} (cm^{-3})$$

The constants A_c and A_v can be derived from the effective density of the states in conduction band N_c (cm⁻³) and valence band N_v (cm⁻³).

$$N_C = A_C T^{3/2}$$
$$N_V = A_V T^{3/2}$$

For silicon at T = 300K (source Pierret): $N_c = 3.22 \times 10^{19} \text{ cm}^{-3}$ and $N_v = 1.83 \times 10^{19} \text{ cm}^{-3}$

$$n_i(T = 300 K) \cong 1 \times 10^{10} cm^{-3}$$

Boltzmann constant = K = $8.617e-5 \text{ eV/K} = 1.38 \times 10^{-23} \text{ J/}^{\circ}\text{K}$ Energy Gap for silicon at room temperature = 1.12 eV

Energy Band Gap

• The energy band gap E_g is affected by temperature according to the following Varshni equation:

$$E_g(T) = E_g(0) - rac{lpha_E T^2}{T+eta_E}$$
 (eV)

- where E_g(0) is the band gap energy at absolute zero and α_E and β_E are material specific constants

Material	$E_g(0)$ (eV)	$\alpha_{\rm E} ~(eV/K)$	$\beta_{\rm E}$ (K)
GaAs	1.519	$5.41*10^{-4}$	204
Si	1.170	$4.73*10^{-4}$	636
Ge	0.7437	$4.77*10^{-4}$	235

Extrinsic Semiconductors (1)

Doping with donor impurities (N-type semiconductor)



Figure 3.3 A silicon crystal doped by a pentavalent element. Each dopant atom donates a free electron and is thus called a donor. The doped semiconductor becomes *n* type.

Extrinsic Semiconductors (2)

Doping with acceptor impurities (P-type semiconductor)



Figure 3.4 A silicon crystal doped with boron, a trivalent impurity. Each dopant atom gives rise to a hole, and the semiconductor becomes

p type.

N-type semiconductor



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electrons

holes

donors

P-type semiconductor (1)



Fig. 2-9 Energy-band diagram of *p*-type semiconductor.

source: Millman & Halkias

source: Howe & Sodini

holes

 $q = 1.6 \times 10^{-19} \, \text{Cb}$

electrons

 $\rho = \text{charge density}[Cb / cm^3] = 0 = (-qn) + (qp) - qN_A$

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acceptors

P-type semiconductor (2)

- Holes can be filled by absorbing free electrons, therefore there is an "effective" flow of holes
- Holes are slower than free electrons (due to the probability of a hole to be filled)
- The effective mass of holes is larger than the effective mass of the free electrons:
 m*_h > m*_e

Mobility of free electrons and holes



Mass Action Law

• The mass-action law is valid for both intrinsic (pure) and extrinsic (doped) semiconductors

$$n_i^2 = n \cdot p$$

 If nî then pl A larger number of free electrons causes the recombination rate of free electrons with holes to increase

Doping with donors (n-type)

• Charge neutrality:

$$\rho = 0 = q(p - n + N_D)$$

• Using mass-action law:

flip sides

$$\frac{n_i^2}{n} - n + N_D = 0 \Leftrightarrow -\frac{n_i^2}{n} + n - N_D = 0 \Leftrightarrow n^2 - N_D \cdot n - n_i^2 = 0$$

$$n = \frac{N_D \pm \sqrt{N_D^2 - 4n_i^2}}{2} \approx N_D \qquad \text{Free electrons} \\ \text{are Majority} \\ \text{Carriers} \qquad p \approx \frac{n_i^2}{N_D} \qquad \text{Holes are Minority} \\ \text{Carriers} \qquad \text{Carriers} \qquad p \approx \frac{n_i^2}{N_D} \qquad \text{Holes are Minority} \\ \text{Carriers} \qquad \text{Carriers} \qquad p \approx \frac{n_i^2}{N_D} \qquad \text{Holes are Minority} \\ \text{Carriers} \qquad \text{Carriers} \qquad p \approx \frac{n_i^2}{N_D} \qquad \text{Holes are Minority} \\ \text{Carriers} \qquad \text{Carriers} \qquad p \approx \frac{n_i^2}{N_D} \qquad \text{Holes are Minority} \\ \text{Carriers} \qquad \text{Carriers} \qquad p \approx \frac{n_i^2}{N_D} \qquad \text{Carriers} \qquad p \approx \frac{n_i^2}{N_D} \qquad \text{Carriers} \qquad p \approx \frac{n_i^2}{N_D} \qquad \text{Holes are Minority} \\ \text{Carriers} \qquad \text{Carriers} \qquad p \approx \frac{n_i^2}{N_D} \qquad \frac{n_i^2}{N_D}$$

Doping with acceptors (p-type)

• Charge neutrality:

$$\rho = 0 = q(p - n + N_D)$$

Doping with N_A >> n_i

$$p \approx N_A$$
 Holes
are Majority
Carriers
 $n \simeq \frac{n_i^2}{N_A}$ Free electrons
are Minority
Carriers

Doping with both donors and acceptors

Charge neutrality:

$$\rho = 0 = q(p - n + N_D - N_A)$$

 Assuming that |N_D-N_A| >> n_i (nearly always true)

For
$$N_D > N_A$$

 $n \simeq N_D - N_A$ and $p \simeq \frac{n_i^2}{N_D - N_A}$

- For N_A > N_D

$$p \simeq N_A - N_D$$
 and $n \simeq \frac{n_i^2}{N_A - N_D}$
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First Carriers Transport Mechanism: Drift



- The process in which charged particles move because of an electric field is called drift.
- Charged particles will move at a velocity that is proportional to the electric field (this is true as long as the field doesn't become too large)



► Figure 2.9 Log-log plot of the drift velocity as a function of electric field for electrons and holes, showing the linear region where the velocity is proportional to the field and velocity saturation at high fields. After R. S. Muller and T. I. Kamins, *Device Electronics for Integrated Circuits*, 2nd ed., Wiley, 1986.



- Eventually the drift velocity saturates: there are too many collisions among `carriers and between carriers and lattice
- v_{sat} for silicon is $\approx 10^7 \text{ cm/s} = 10^5 \text{ m/s}$



Drift current and current density

• electric current: amount of charge that flows through a reference plane per unit time



Drift Current Density



Conversion between resistivity and dopant density of silicon at room temperature



Second Carriers Transport Mechanism: Diffusion

 The thermal motion of an electron or a hole changes direction frequently by scattering off imperfections in the semiconductor crystal



source: Streetman

> In a material where the concentration of particles is uniform the random motion balances out and no net movement result (drunk sail-man walk → Brownian walks)

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Random thermal motion of an electron or hole in a solid.



Diffusion Current

- If there is a difference (gradient) in concentration between two parts of a material, statistically there will be more particles crossing from the side with higher concentration to the side with lower concentration than vice versa
- Therefore we expect a net flux of particles





The more non uniform is the concentration the more is the current

Electron and hole diffusion current

 Assuming the charge concentration decreases with increasing x it means that dn/dx and dp/dx are negative quantities so to conform with conventions we have to put a – sign in front of the proportionality constant D

$$I_n^{diff} = -D_n A q_e \frac{dn}{dx} = D_n A q \frac{dn}{dx}$$

$$I_p^{aug} = -D_p A q_p \frac{1}{dx} = -D_p A q \frac{1}{dx}$$





Source: Howe & Sodini

Diffusion current densities

$$J_{diff} = J_p^{diff} + J_n^{diff}$$

$$J_n^{diff} = -D_n q_e \frac{dn}{dx} = D_n q \frac{dn}{dx}$$
$$J_p^{diff} = -D_p q_p \frac{dp}{dx} = -D_p q \frac{dp}{dx}$$





Source: Howe & Sodini

- X

Einstein's Relation

 Since both µ and D are manifestation of thermal random motion (i.e. are due to statistical thermodynamics phenomena) they are not independent

$$\frac{D_p}{\mu_p} = \frac{D_n}{\mu_n} = \frac{KT}{q} \quad \longleftarrow \quad \text{Einstein's Relation}$$

K=Boltzmann Constant = 1.38×10^{-23} J/°K = 8.62×10^{-5} eV/°K T = temperature in °K q = charge of proton = 1.602×10^{-19} Cb

$$V_T \triangleq \frac{KT}{q} \longleftarrow$$
 Thermal Voltage

At room temperature $V_T \approx 25.9 \text{ mV}$

Total current density

• The electron and hole total current density is:

$$J = J_p + J_n = J_p^{drift} + J_p^{diff} + J_n^{drift} + J_n^{diff}$$

$$J_{p} = J_{p}^{drift} + J_{p}^{diff} = qp\mu_{p}E - qD_{p}\frac{dp}{dx}$$

$$J_n = J_n^{drift} + J_n^{diff} = qn\mu_n E + qD_n \frac{dn}{dx}$$

$$J = qp\mu_p E - qD_p \frac{dp}{dx} + qn\mu_n E + qD_n \frac{dn}{dx}$$